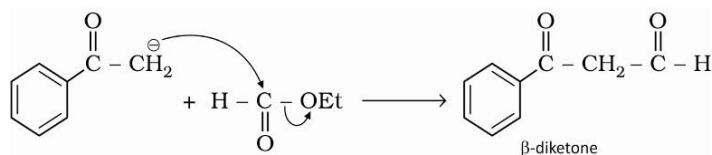
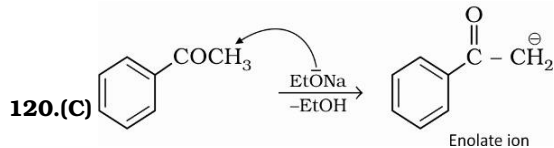
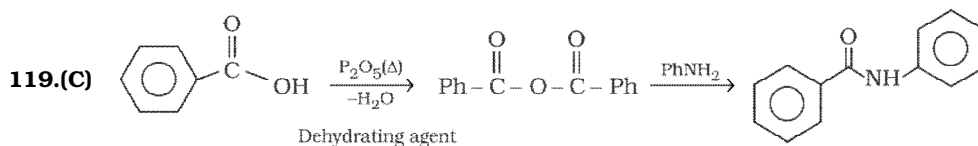
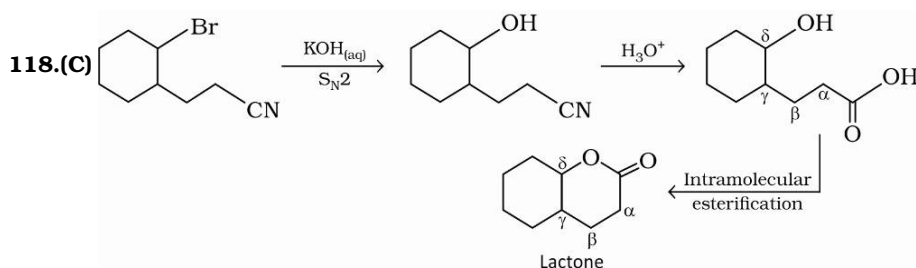
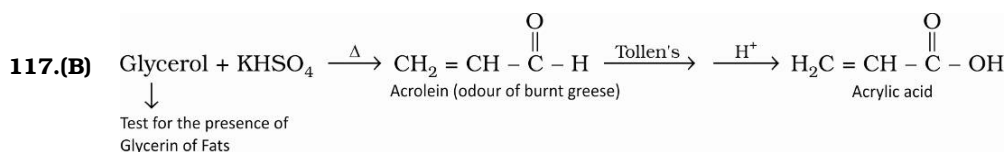
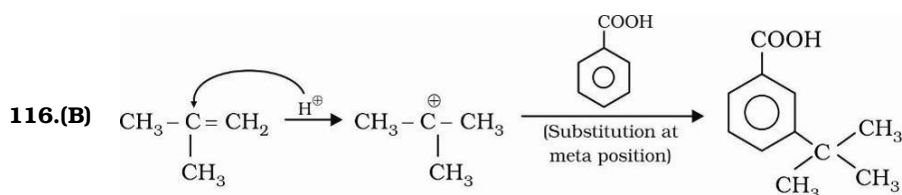
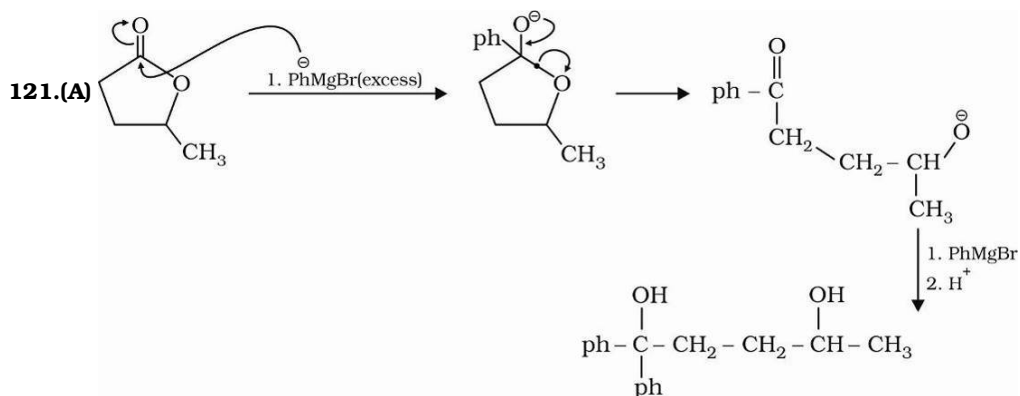


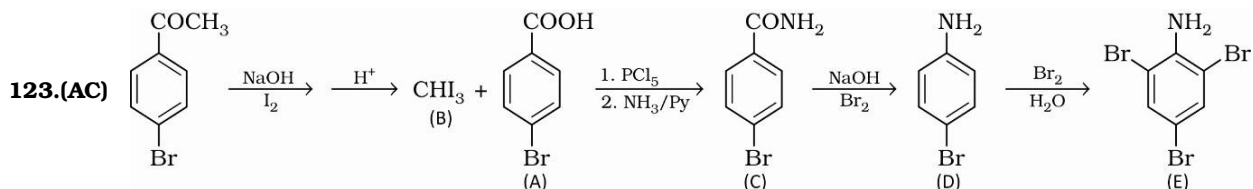
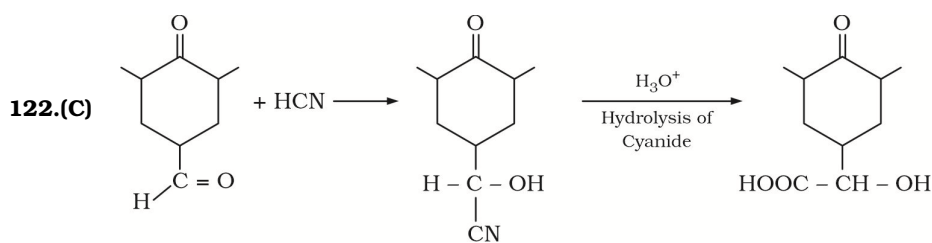
Daily Tutorial Sheet-10

Level-2



(Visualise Claisen type condensation where one species loses α-H and other loses alkoxide)





124.(C) NaHCO_3 only reacts with acid stronger than H_2CO_3
 $\text{H}-\text{COOH}$ is stronger than H_2CO_3 and $\text{H}-\text{CHO}$ is not acidic.

